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This is in accordance with the second law of thermodynamics that a spontaneous process leads to increase in entropy if the system. Step 2 of 2 On the other hand, separation is not a spontaneous process and requires energy, hence is positive. Separation means de-mixing of the components, thus creating more phases.

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By the second law of thermodynamics, a natural process tends to randomness. The separation of a mixture does not occur naturally or spontaneously. Energy is required to separate the different molecular species. Exercise 1.3 Subject: Separation of a mixture requires a transfer of energy to it or the degradation of its energy.

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