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valence electrons. Step 2: Write the skeleton structure of the molecule. Step 3: Use two valence electrons to form each bond in the skeleton structure. Step 4: Try to satisfy the octets of the atoms by distributing the ...

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States that matter is composed of extremely small particles called atoms; atoms are invisible & indestructible; atoms of a given element are identical in size, mass, & chemical properties; atoms of a specific element are different from those of another element; different atoms combine in simple whole-number ratios to form compounds; in a chemical reaction, atoms are separated, combined, or rearranged.

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whenever two elements form more than  
one compound, the different masses of one  
element that combine with the same mass  
of the other element are in the ratio of

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small whole numbers. daltons atomic theory. 1) all matter is composed of atoms. 2)atoms of an element are identical in size mass and other properties.

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